KULEV, L.P.; VGRONOVA, K.R.

Azo deriva ives of barbituric and thiobarbituric acids and their physiological activity. Izv.TPI 111:30-35 '61. (MIRA 16:9)
(Barbituric acid--Physiological effect)
(Azo compounds)

KULEV, L.P.; STEPNOVA, G.M.; TABINSKAYA, P.F.

Substituted amides of 4-nitro-2,2'-diphenic acid. Izv. TPI 126;51-52
'64. (MIRA 18:7)

KULEV, L.P.; GIREVA, R.N.; HELYAYEVA, A.P.

Diphenic acid esters. Part 4: Monoaryl esters of diphenic acid and their insecticide activity. Izv. TPI 126:53-54 '64. (MIRA 18:7)

"APPROVED FOR RELEASE: 08/23/2000

KULEV. N.: BEZZUEENKO. B.

Modernization of the drive of the automatic hamburger shaper. Mias.ind. SSSR 33 no.3:46-47 162. (MIRA 25:7) SSSR 33 no.3:46-47 162.

1. Semipalatinskiy myasokombinat. (Meat industry—Equipment and supplies)

#### KULEV, N.

Fastening of the blades of FMSh-340 and FMSh-650 fill mixers. Mias.-ind. SSSR 33 no.3:48-49 '62. (MIRA 15:7)

1. Semipalatinskiy myasokombinat.
(Mixers (Machinery)) (Mat industry—Equipment and supplies)

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R000927410015-2"

S/135/61/000/009/006/006 A006/A101

AUTHOR:

Kulev, Yu. I., Engineer

TITLE:

One-sided argon-arc welding of thin sheet steel with least rein-

forcement of the weld on the reverse side

PERIODICAL:

Svarochnoye proizvodstvo, no. 9, 1961, 27

To produce weld joints without reinforcement on the reverse side, the author used argon-arc welding with non-consumable electrode, with or without filler wire and one or two high-frequency (480 cycles) NC-100 (PS-100) transformers. Structures of 3 and 2.5 mm thick 25 XTC A(25KhGSA) and 30 XTCA (30KhGSA) steel were welded. When employing one PS-100 transformer for welding 3 mm thick 25KhOSA steel at 10 m/hr speed, 115 - 120 amps current, and 10 m/hr filler wire feed, an up to 0.4 mm deep meniscus was formed on the internal side of the weld joint. When using two parallel connected transformers for welding 3 mm thick 25KhGSA steel at 165 - 175 amps current, 16 m/hr welding speed, 16 m/hr filler wire feed and 3 mm diameter tungsten electrode, the reinforcement on the reverse side was 0 - 0.2 mm thick, and 0.3 - 0.6 mm on the external side of the weld. When producing weld joints without filler wire at 115 - 120 amps

Card 1/2

CIA-RDP86-00513R000927410015-2"

**APPROVED FOR RELEASE: 08/23/2000** 

One-sided argon-arc welding ...

S/135/61/000/009/006/006 A006/A101

current and 13 - 15 m/hr welding speed, using 2 mm diameter tungsten electrodes, a weld joint with a 0.1 mm meniscus on the internal side was obtained. The use of a copper backing plate did not affect the thickness of the reinforcement on the reverse side. The welding process with the use of a PS-100 transformer is characterized by a large penetration width of the butt on the internal side which is equal to about 70% of the external weld width, and by a smaller molten pool than in welding on d-c or a-c of power frequency. There are 3 figures.

Card 2/2

MULETA, A.K.

34027 <u>KULEYA, A.K.</u> Normalizatsiya Zapravok V Lyentotkachyestvye Lyegkaya Prom-st; 1949, No. 9 S. 22-24

SO: Letopis'Zhurnal'nykh Statey, Vol. 42, Moskva, 1949

#### "APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R000927410015-2

KULEVA, A.K.

Galculating the consumption of rubber filaments. Leg. prom. 17 no.12: 32-35 D '57. (MIRA 11:1)

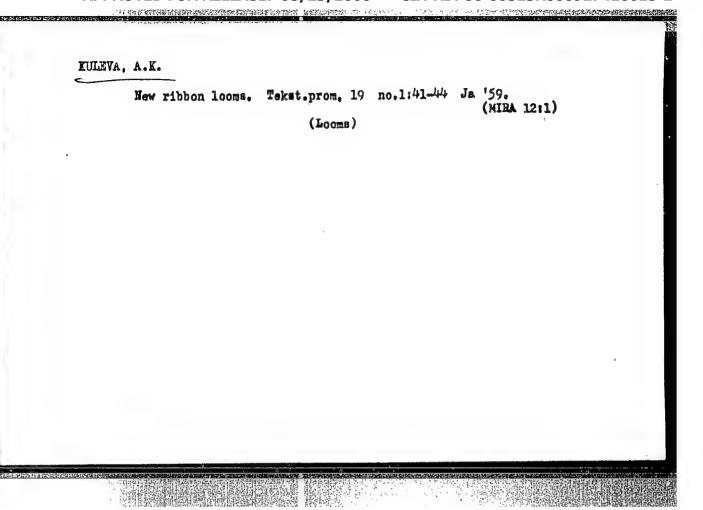
(Rubber goods)

IVANOV, V.A., kand.tekhn.nauk; KULEVA, A.K., inzh.; RYNSKIY, G.V., inzh.

Synthetic substitutes for metals. Leg.prom. 18 no.12:35-37
D '58.

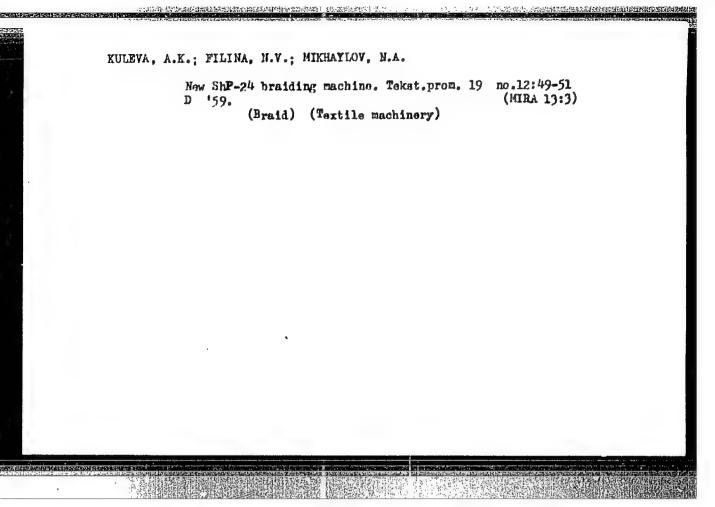
(Metals, Substitutes for)

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R000927410015-2"



New ShP-52 and LP-53 braiding machines. Tekst.prom. 19 no.2:
40-42 F '59. (MIRA 12:5)

(Textile machinery)

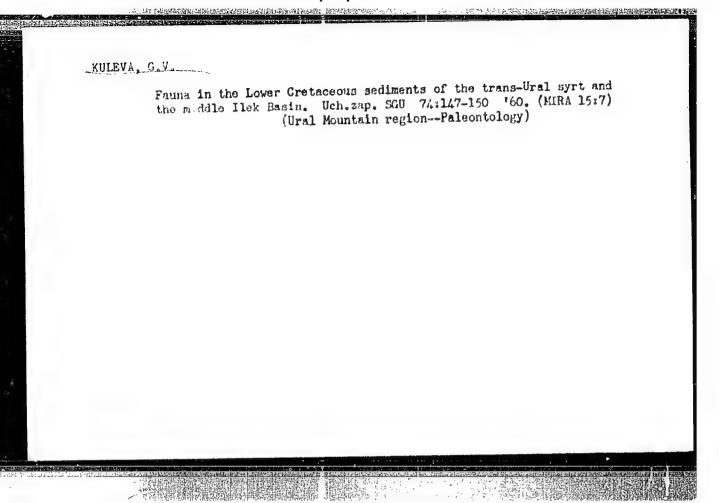


KULEVA, A.K.

Mochanization and automation of the basic and auxiliary operations in the ribbon weaving industry. Tekst. prom. 24 no.2:56-58 F '64. (MIRA 17:3)

1. Zaveduyushchiy tekhnologicheskim otdelom Vsesoyuznogo nauchno-issledovateliskim institutom tekstilino-galantereynoy promyshlenosti.

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R000927410015-2"



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#### "APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R000927410015-2

TSUPAK, Valerian Fedorovich, kand.sel'skokhosyayatvannykh nauk; KULEVA, Iraida Fedorovna, kand.sel'skokhosyayatvannykh nauk; SINYAKOVA, Lidiya Andreyevna, kand.biol.nauk; VOROB'YEV, F.I., red.; CHUMAYEVA, Z.V., tekhn.red.

[Practical laboratory experiments in plant culture] Laboratorno-prekticheskie zaniatiia po rastenievodstvu. Moskva, Gos. izd-vo sel'khos. lit-ry, 1957. 255 p.

(Plants, Cultivated)

KULEVA, I.F., dots., kand. sel'khoz. nauk; MAL'CHIKOVA, V.K., red.;
TIKHONOVA, I.M., tekhn. red.

[Sugar beets; practices in growing sugar beets in Leningrad-Province] Opyt vozdelyvaniia sakharnoi svekly v Leningradskoi oblasti. Leningrad, Lenizdat, 1963. 113 p. (MIRA 16:10)

(Leningrad Province-Sugar beets)

KNYAZEVA, T.S.; KORSHAK, V.V.; AKUTIN, M.S.; KULEVA, M.M.; VINOGRADOVA, S.V.; RODIVILOVA, L.A.; NEDOPEKINA, T.P.; VALETSEIT, P.M.; MOROZOVA, S.A.; SALAZKIN, S.N.

Possibility of using various polyarylates as insulating film (MIRA 16:1) (Acids, Organic) (Polymers) (Insulating materials)

C N1	T) L 11235-66 EVIT (m) /EWP (1) /T/EWA (c) /ETC (m) /W/RM  SOURCE CODE: UR/0080/65/038/012/2728/2734  WH 55 44 55	;
UTHOI	R: Koton, M. M.; Yakovlev, B. I.; Rudakov, A. P. Knyazeva, T. S.; Florinskiy, 73; Bessonov, M. I.; Kuleva, M. H.; Tolparova, G. A.; Layus, L. A.	r
RG: loyed	ineniy AN SSSR) 44 55	,
TTLE	: Preparation and physicomechanical properties of polypyromellitimide	
ourc	E: Zhurnal prikladnoy khimii, v. 38, no. 12, 1965, 2728-2734	;
OPIC olyi	TAGS: heat resistant plastic, fire resistant material, dielectric material, mide, polypyromellitimide/material	
prope the p mater	CACT: A study has been made of the preparation and physical and mechanical erties of a polyimide, viz., polypyromellitimide. Test results showed that polymer may find widespread use as a heat resistant and low temperature resistant rial, and is of special interest as a high temperature film dielectric. A polymellitimide film similar to the U.S. H-film was prepared from pyromellitic and the description of the control of the	•
,	$0 < \frac{\text{CO}}{\text{CO}} R < \frac{\text{CO}}{\text{CO}} O + \text{II}_{2}N - \text{II} - \text{NII}_{3} \rightarrow \begin{bmatrix} \text{IIO} - \text{CO} \\ -\text{NII} - \text{CO} \end{bmatrix} R < \frac{\text{CO} - \text{OII}}{\text{CO} - \text{NII} - \text{R'}} = \begin{bmatrix} \text{IIO} - \text{CO} \\ -\text{NII} - \text{CO} \end{bmatrix} R$	:
Card	$- + \left[ -N \left\langle {^{CO}_{CO}} R \left\langle {^{CO}_{CO}} N - R' - \right  \right]_{*},  \text{UDC};  .541.6$	!

THE PROPERTY AND THE PROPERTY AND THE PROPERTY OF THE PROPERTY

I. 11235-66

ACC NRI AP6002214

Polycondensation to the polyamido acid intermediate was carried out at 15C. Poly-pyromellitimide films were prepared by drying solutions of the polyamido acid on glass substrates at 20—40C followed by heat treatment at 80—400C to produce imidization. Optimum preparative conditions were determined. The films were transparent, gold-brown in color, thermally stable, nonburning at up to 600—700C, unaffected by organic solvents, highly resistant to \( \gamma \) and UV radiation, low temperature resistant, nonshrinking, resistant to humidity, and readily metalized. In its mechanical properties at high temperatures, the material surpasses all existing polymers. These properties can be further improved by orientation stretching, after which they approach those of glass-reinforced plastics and metals. Orig. art. has:

SUB CODE: 11/ SUBM DATE: 08Mar65/ ORIG REF: 008/ OTH REF: 011/

ATD PRESS: 4/73

Cord 2/2

PONOMAREV-STEPNOT, N. N.; SMIRNOV, O. N.; KULEVA, R. V.

"Investigation on System with Zirconium Hydride Moderator."

report submitted for 3rd Intl Conf on the Peaceful Uses of Atomic Energy, Geneva, 31 Aug-9 Sep 64.

IJP(c) EMT(m)/EWP(t)/ETI SOURCE CODE: UR/0075/66/021/001/0046/0052 L 36081-66 (A) ACC NRI AP6016300 30 Kuleva, V. M.; Popovs, A. N. B AUTHOR: ORG: Leningrad Pedagogical Institute im. A. I. Gertsen (Leningradskiy pedagogicheskiy institut) TITLE: Use of o-nitrobenzoic scid for quantitative determination of zirconium and hafnium Zhurnal analiticheskoy khimii, v. 21, no. 1, 1966, 46-52 SOURCE: TOPIC TAGS: hafnium, zirconium, quantitative analysis ABSTRACT: The starting solutions of zirconium were prepared from zirconium chloride twice recrystallized from hydrochloric acid. The hafnium solutions were prepared from "experimental" grade hafnium hydroxide which was then dissolved in concentrated hydrochloric acid. In the precipitation of zirconium or hafnium the procedure was as follows. To a determined volume of a standard solution of zirconium or hafnium salt there was added a corresponding amount of concentrated hydrochloric acid to create the required acidity, and the solution was heated to the start of boiling. For each 100 ml of the final solution there was added about 5 grams of ammonium nitrate or obloride. Then, UDC: 543.70 Card 1/2

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R000927410015-2"

The same of the sa

L 36081-66

ACC NR: AP6016300

a hot 1.5% solution of o-nitrobenzoic acid was added with constant stirring. The precipitate was filtered and calcined to constant weight in a mufflo furnace. The experimental results are presented in a series of figures and tables. It is shown that, for selectivity and sensitivity, o-nitrobenzoic acid is not inferior to m-nitrobenzoic acid, and that it has the advantage of a greater solubility in water and a higher ionization constant. The composition of the zirconium and hafnium o-nitrobenzoates precipitated depends on the acidity of the medium. Ti(IV), Ti(III), Sn(IV) and Sn(II) interfere with the precipitation process. In the presence of Th4+, Fe<sup>3+</sup>, and Cr<sup>3+</sup> reprecipitation is necessary. Orig. art. has: 3 figures and 6 tables.

SUB CODE: 07/ SUBM DATE: 22Dec64/ ORIG REF: 007/ OTH REF: 015

Card 2/2

KULEVA , V.N.; POPOVA, A.N.

Use of o-nitrobenzoic acid for the quantitative determination of zirconium and hefnium. Zhur. anal. khim. 21 no. 1\*46-52 '66 (MIRA 19:1)

1. Lenlngradskiy pedagogicheskiy institut imeni Gertsena.

RUDWARUT, S.I.

Boots and Shoes - Trade and Manufacture

Construction of uppers of the shoe "Parko 2." Leg. prom. 12 no. 4:45 Ap 152

Monthly List of Russian Accessions, Library of Congress, July 1952. Unclassified

## KULEVATSKIY, S.

Expanding ticket sale by booking offices. Avt. transp. 41 no.6:18 Je '63. (MIRA 16:8)

1. Zamestitel' upravlyayushchego Tambovskim oblastnym avtomobil'nym trestom po passazhirskim perevozkam.

KULEVOV, A., starshiy pilot-inspektor

Some problems in methodology. Grazhd. av. 21 no.10:22 0 \*64.

(MIRA 18:3)

1. Neshtatnyy korrespondent zhurnala "Grazhdanskaya aviatsiya."

KUKOLEVSKIY, Georgiy Mikhaylovich; KULEVOVA, A.M., red.; GABERLAND, M.I., tekhn.red.

[Doctor's advice to the athlete] Sovety vracha sportsmenu.

Moskva, Gos.ixd-vo med.lit-ry, 1958, 188 p. (MIRA 13;4)

(PHYSICAL EDUCATION AND TRAINING-HYGIENIC ASPECTS)

YEROFEYEV, B.V.; NAUMOVA, S.F.; KULEVSKAYA, I.V.

Initiation of ethylene polymerization by the action of a complex consisting of etherates of Griggard compounds and titanium tetrachloride. Shor. nauch. rab. Inst. fiz.-org. khim. AN BSSR no.8:80-82 '60.

(MIRA 14:3)

1. Institut fiziko-organicheskoy khimii AN BSSR.
(Ethylene) (Polymerization) (Titanium compounds)

5/190/61/003/011/012/016 B110/B101

Yerofeyev, B. V., Naumova S. F., Kulevskaya, I. V. Mardykin AUTHORS:

V P Tsykalo, L G,

Polymerization of ethylene in the presence of the triethyl TITLE:

aluminum anisolate and titanium tetrachloride complex

PERIODICAL: Vysokomolekulyarnyye soyedinentya, v 3, no 11, 1961, 1705

- 1707

TEXT: Initiators from triethyl aluminum anisolate (A) and TiCl (T) for ethylene polymerization have low self-inflammability. The authors studied the properties of polyethylene (PE) produced with them, and the effect of the A:T ratio on its properties. The A1(C2H5)3 CH30C6H5 was prepared by reaction of bromo ethyl with Mg-Al alloys (40% Al; 60% Mg in anisole).
1.0 mole/liter of A (boiling point 97 - 105 C/4-5 mm Hg) was dissolved in heptane. The TiCl<sub>4</sub> concentration in n-heptane was 0.4 moles/liter. Ethylene was pressed into the reaction vessel at 12 liters/hr. At first n heptane, after this TiCl in n-heptane, and then during 1 min, A in Card :/3

CIA-RDP86-00513R000927410015-2" APPROVED FOR RELEASE: 08/23/2000

S/190/6:/003/011/012/019 B:10/B101

Polymerization of ethylene in the...

n-heptane were added. After 20 min, PE was precipitated by means of The tabulated values were found under atmospheric CH\_OH with 5% HCT pressure at 30°C. The density determined in water-alcohol mixture was 0.99 0 97. With increasing A:T ratio and constant T, the molecular waight of PE drops. Then, the amount of A determines the number of resulting polymer macromolecule chains. The A:T ratio was < 1 in teste . . 5 and 1.5 in test 6. While PE obtained by means of triisobutyl aluminum and TiCl4 (Ref. 5, see below) had molecular weights of 67 000 940,000 and melting temperatures of 116 - 139°C, the molecular weights of the authors! PE were 91,000 - 316,000, the melting temperatures 127 150 C. The recrease of the molecular weight with decreasing Al compound: TiCl ratio observed in triisobutyl aluminum polymer ration is probably Thus the TiCl, amoun' determines ing to the high extess of the former There are I table and 5 non-Series the number of resulting polymer chains references. The two references to English-language publications resolut Tablews. Ref 2: A Grosse J Mavi'y, J Org Chem. 5, 106, 1940; Raf hi E Badin J Amer Chem Ber 30 Care and

Card 2/5

Folymerication of ethylene in the...

5/190/61/003/011/012/016 B110/B101

ASSOCIATION: Institut fizikoorganicheskoy khimii AN BSSR (Institute of Physical and Organic Chemistry AS BSSR)

SUBMITTED:

December 26, 1960

Table. Ethylene polymerization.

Legend: (;) test no.; (2) amount of initiator components; (3) millimoles; (4) polyethylene yield, g; (5) molecular weight; (6) melting point, C.

Omat	<ol> <li>Колнчество номпонентов инициатора</li> </ol>			<b>(</b>	(5)	
.\;	3 MMOSTU	3 M.MO.TU	A/T	Выход полиэтилена, *	Молекулярны <b>а</b> вес	T. n.a., *C
1 2 3 4 5	1,23 2,47 3,70 3,51 4,95 6,57	6,0 6,0 6,0 5,0 6,0 4,0	0,21 0,41 0,62 0,70 0,82 1,64	1,37 1,98 2,60 2,34 2,89 2,52	316 000 250 000 180 000 91 000	128 130 127 130

Table

Card 3/3

S/786/61/000/009/001/006 1065/1242

AUTHORS: B.V. Yerofeyev, S.F. Naumová, V.P. Markykin, I.V. Kulevskaya,

L.G. Taykalo

TITLE: The dependence of the molecular weight of polyethylene

on the TiCl /Al(iso-C, Hq), ratio in the Ziegler catalyst

SOURCE: Akademiya nauk Belorusskoy SSR. Institut fiziko-organi-

cheskoy khimii. Sbornik nauchnykh rabot. no.9. 1961. Monomery, svoystva i protsessy polucheniya polimerov.

59-62

TEXT: In the polymerization of ethylene initiated by a Ziegler catalyst with excess TiCl<sub>4</sub>, the molecular weight of the polyethylene obtained increases with decrease of the [AlR<sub>3</sub>]/[TiCl<sub>4</sub>] ratio. These findings disagree with the data of Badin (J.Am.Chem.Soc. 80, 6545, 1958). The polymerizations were carried out in a glass vessel equipped with mechanical stirrer, reflux condenser, gas inlet tube and a burette for the introduction of the dissolved catalyst components. Molecular weights were determined viscometrically

Card 1/3

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The dependence of the molecular ...

S/786/61/000/009/001/006

1065/1242

The interpretation of the experimental interpretation of a very high value for the subscript of the complex initial concentration in the subscript of the TiGl4 - AlR3 complex can be represented by concentration (X) of the TiGl4 - AlR3 complex can be represented the approximate expressions [TiGl4] of [AlR3] of [
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# APPROVED FOR RELEASE: 08/23/2000

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The dependence of the molecular ... $\frac{5/786}{61}/000/009/001/006}$$ weight of polyethylene will increase on decreasing the [AlR3]/[TiCl4] ratio when [AlR3] < [TiCl4] or or on increasing the [AlR3]/[TiCl4] ratio when [AlR3] < [TiCl4] or There are 3 tables.
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• 17、大型的化量制度和电影的影響的影響的影響的影響的影響的影響的影響。

L 33526-66 EWT(m)/T/EWP(j) IJP(c) WW/RM
ACC NRI AP6015052 (A) SOURCE CODE: UR/0190/66/0 9/005/0876/0881
AUTHOR: Kulevskaya, I. V.; Yerusalimskiy, B. L.; Mazurek, V. V.
ORG: Institute of Macromolecular Compounds, AN SSSR (Institut vysokomolekularnykh soyedineniy AN SSSR)  TITLE: Polymerization kinetics of the acrylonitrile under the effect of butylmagnesium chloride
SOURCE: Vysokomolekulyarnyye soyedineniya, v. 8, no. 5, 1966, 876-881
NOPIC TAGS: polymer, monomer, polymerization kinetics, acrylonitrile, magnesium, chloride, toluene, macuesum compound
ABSTRACT: The kinetics of polymerization in the system acrylonitrile, toluene, and outylmagnesium chloride at -75C has been investigated. A mechanism of the colymerization process involving elementary stages through intermediate complexes was proposed. For the initial stage of polymerization, the first order of the catalyst and the second order of the monomer were shown. The molecular weight of the polymers exceeded 200,000. Orig. art. has: 7 figures, 9 formulas, and 1 table.
UB CODE: 07/ SUBM DATE: 13May65/ ORIG REF: 006/ OTH REF: 012
Gard 1/1 90. UDC: 66.095.26+678.745

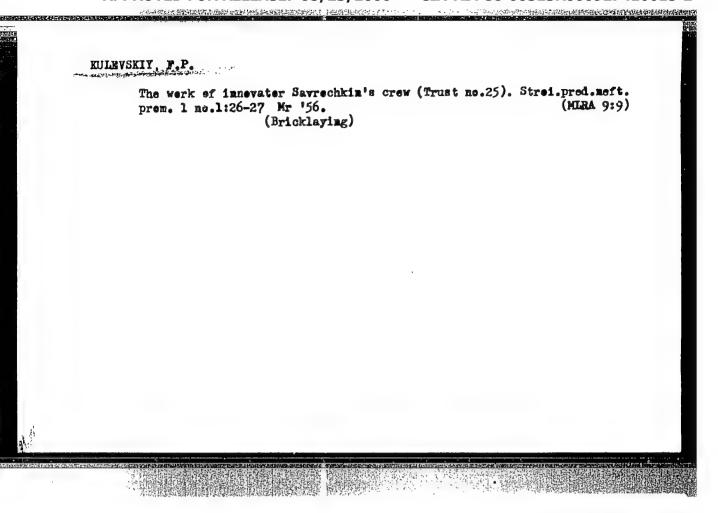
### "APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R000927410015-2

37646-66 EWT(m)/EWP(j)/TACC NRI AP6011237 (A) SOURCE CODE: UR/0413/66/000/006/0075/0075 34 INVENTOR: Yerusalimskiy, B. L.; Kulevskaya, I. V.; Kamalov, S. K.; B Frenkel', S. Ya. ORG: none TITLE: Preparation of polyacrylonitrile. Class 39, No. 179925 [announced by the <u>Institute of High-Molecular Compounds</u>, AN SSSR (Institut vysokomolekulyarnykh soyedineniy AN SSSR)] SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 6, 1966, 75 TOPIC TAGS: polyacrylonitrile, acrylonitrile, polymerization ABSTRACT: This Author Certificate introduces a method of preparing polyacylonitrile by polymerization of acrylonitrile in a hydrocarbon solvent at about -750 in the presence of grand agnesium catalysts. To extend the variety of organomagnesium cataly as, complexes of magnesium alkyl halides or magnesium alkyls wit. !imethylsulfoxide are suggested. [LD] 11.07/ SUBM DATE: 13F0b65 SUB CODE: Card 1/1 Amp UDC: 678.745.32

YERUSALIMSKIY, B.L.; KULEVSKAYA, I.V.

Polymerization of acrylonitrile under the effect of organomagnesium compounds. Vysokom.soed. 7 no.1:184-185 Ja '65. (MIRA 18:5)



Kuleuskiy L.A. Wald nr. 971-13 20 May STUDY OF RUBY LASER AT LIQUID NITROGEN TEMPERATURE (USSR)

Konyukhov, V. K., L. A. Kulevskiy, and A. M. Prokhorov. IN: Akademiya hauk SSSR. Doklady, v. 149, no. 3, 21 Mar 1963, 571-572.

S/020/63/149/003/012/028

Spectral components of ruby laser emission corresponding to laser transitions to the  $\pm 1/2$  and  $\pm 3/2$  components of the ground state have been studied at 77.4°K, A light-pink ruby sample 6 mm in diameter and 60 mm long was used, with one end silver-coated and the other uncoated. The laser beam was passed through a Fabry-Perot interferometer with a 0.20-cm air gap into a long-focus camera, where it was either photographed on red-sensitive film or separated into the two components by a mask. In the latter case each component was detected separately by a photomultiplier, and the two signals were registered by a dual-beam oscillograph. Near the laser threshold only the  $\pm 3/2$  (short-wave) component was observed, the other appearing at higher pumping energies. The frequency difference of the two components, calculated from the interference pattern  $(0.36 \pm 0.03)$  cm<sup>-1</sup> agrees, within the experimental error, with a value calculated from the splitting of the  $Cr^{5+}$  ground state in the  $Al_2O_5$  lattice (the ground state being determined by EPR methods). It was determined

Card 1/2

AID Nr. 971-13 20 May

STUDY OF RUBY [Cont'd]

8/020/63/149/003/012/028

that the components carry different fractions of the output energy: near the threshold the short-wave component carries most of the energy, while the long-wave component increases to  $21 \pm 1\%$  of the short-wave component considerably above the threshold. The time variation of the two components was shown to be quite dissimilar. The short-wave component was generated in 0.5 to 0.8  $\mu$ sec, and its duration increased with increased pumping energy; the long-wave component was generated in 0.1 to 0.15  $\mu$ sec, and its duration decreased with increased pumping energy. [BB]

Card 2/2

KONYUKOV, V.K.; KULEVSKIY, L.A.; PROKHOROV, A.M.

Ruby-operated laser with a generation length of ~10 msec. Zhur. eksp. i teor. fiz. 45 no.4:857-862 0 '63. (MIRA 16:11)

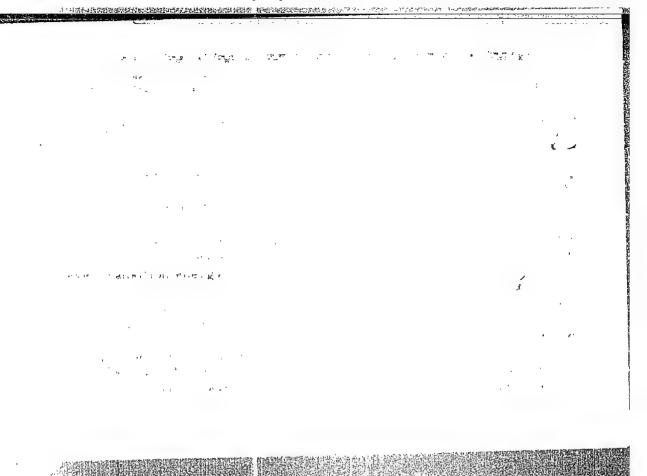
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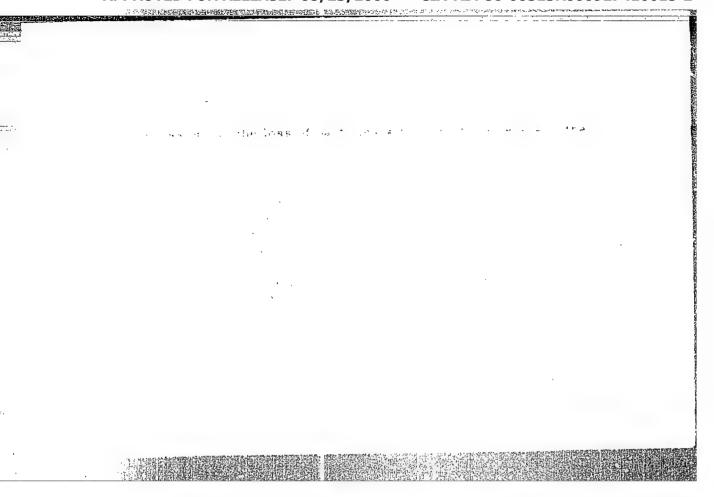
1. Fizicheskiy institut imeni P.N.Lebedeva AN SSSR.

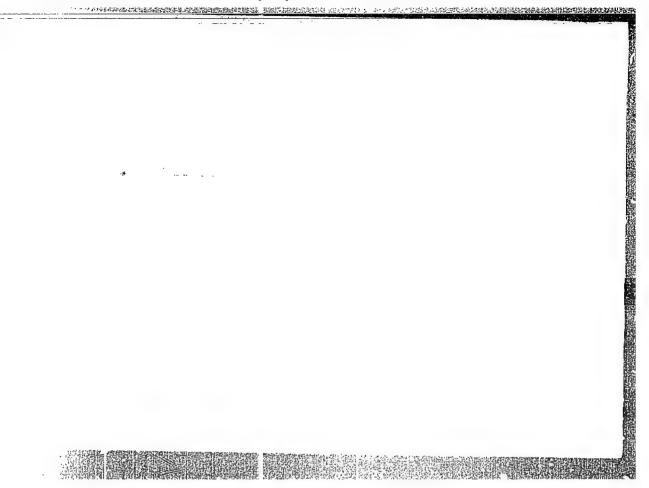
KONYUKHOV, V.K.; KULEVSKIY, L.A.; PROKHOROV, A.M.

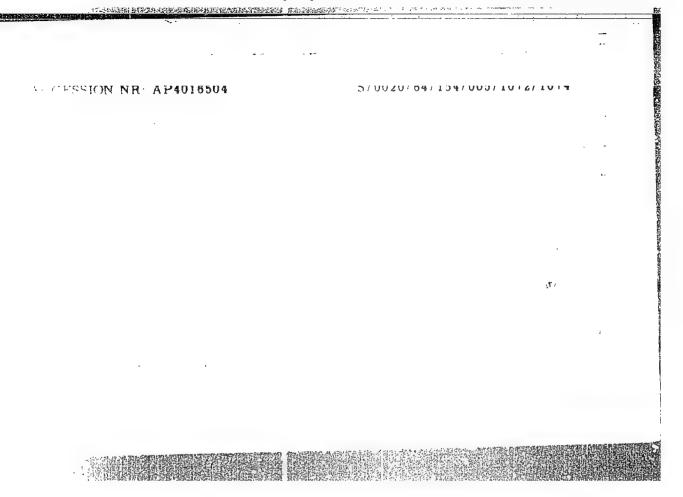
Internal oscillation types in a ruby laser. Dokl. AN SSSR 154 no.5:1072-1074 F'64. (MIRA 17:2)

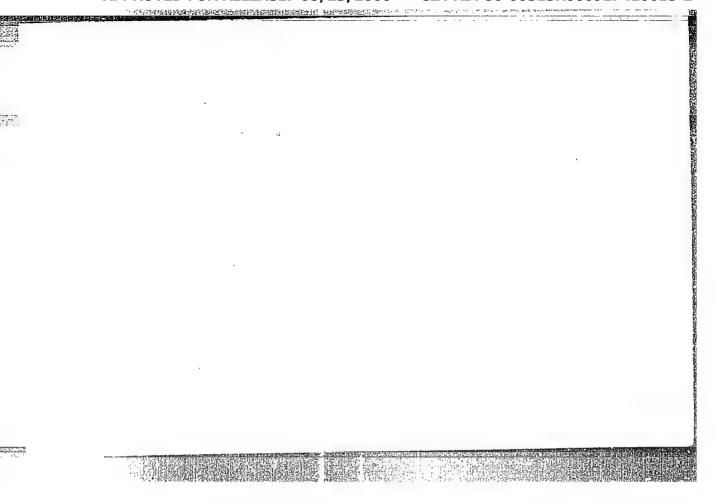
1. Fizicheskiy institut im. P.N. Lebedeva AN SSSR. 2. Chlen-korrespondent AN SSSR (for Prokhorov).

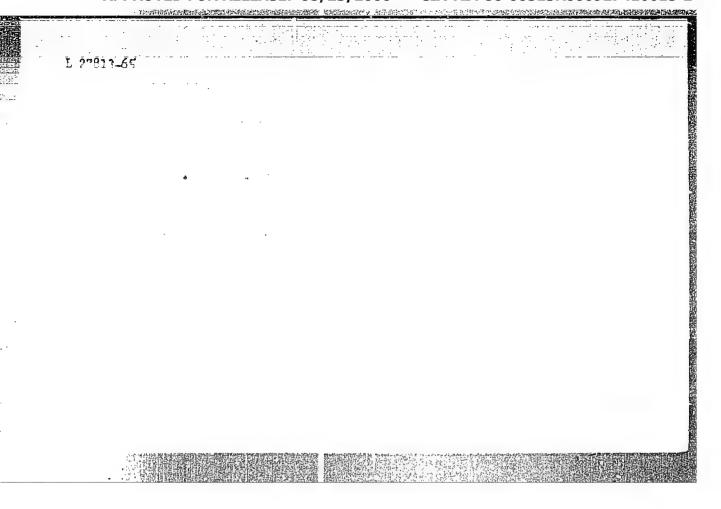












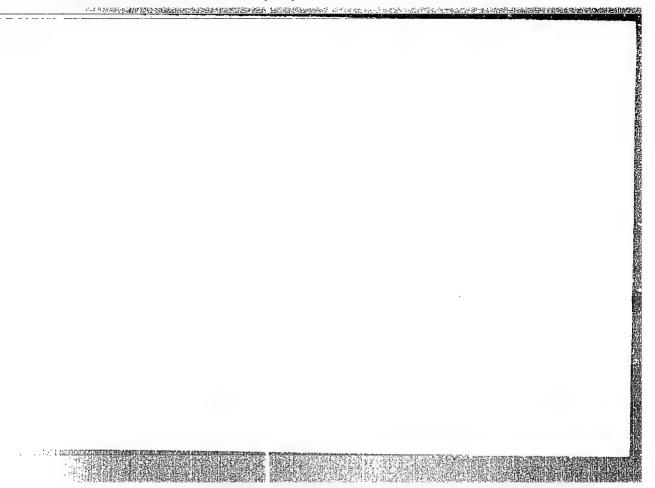
KONYUKHOV, V.K.; KULEVSKIY, L.A.; PROKHOROV, A.M.; SOKOLOV, A.K.

Spectrum of a ruby optical maser with external spherical mirrors.

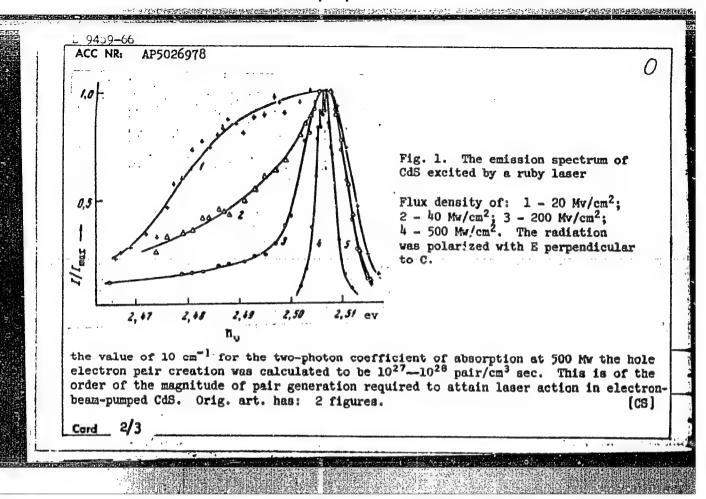
Dokl. AN SSSR 158 no.4:824-826 0 164. (MIRA 17:11)

1. Fizicheskiy institut im. P.N. Lebedeva AN SSSR. 2. Chlen-korrespondent AN SSSR (for Prokhorov).





ACC NR. AP5026978 SCTB/IJP(c) WG/JD/WH SOURCE CODE: UR/0020/65/164/005/1012/1012  AUTHOR: Konyukhov, V. K.; Kulevskiy, L. A.; Prokhorov, A. M. W/  ORG: Physics Institute im. P. N. Lebedey, Academy of Sciences SSSR (Fizicheskiy institut Akademii nauk SSSR)  TITLE: A cadmium sulfide laser using two-photon excitation from a ruby laser  SOURCE: AN SSSR. Doklady, v. 164, no. 5, 1965, 1012-1015  TOPIC TAGS: laser, semiconductor laser, nonlinear optics, two photon absorption  ABSTRACT: A CdS 5 x 3 x 3 mm laser forming a Fabry-Perot cavity was excited by focused radiation from a 50 Mw Q-switched ruby laser. The emission spectrum of CdS-was investigated at flux densities of 20, 40, 200, and 500 Mw/cm² with laser action ference pattern was obtained and a beam directivity of v10° above the threshold was of the exciting light at all excitation levels; however, the duration of the bell-the fact that the power absorbed shows a quadratic dependence on the incident power. at 300K and found to be proportional to the flux density of incident radiation (0.2, 0.5, and 1.1 cm at flux densities of 10, 25, and 55 Mw, respectively). Using Cord
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L 10949-66 FRD/FHT(1)/FHP(a)/EVT(m)/FEC(k)-2/T/EVP(t)/EVP(k)/EVP(t)/EVA(m)  ACC NR: AP6002423 SCTR/IJP(c) SOURCE CODE: UR/0020/65/165/005/16  WEJ/JD/WH  AUTHOR: Konyukhov, V. K.; Kulevskiy L. A.; Kostin, V. V.; Murina, T. M.; P. A. M. (Corresponding member AN SSSR)	-2/EWA(h) 56/1058 9/ rokhorov,
ORG: Physics Institute im. P. N. Lebedev Academy of Sciences, SSSR (Fiziche institut Akademii nauk SSSR)  TITLE: A giant-pulse CaF <sub>2</sub> : Dy <sup>2+</sup> laser with continuous pumping	skiy
SOURCE: AN SSSR. Doklady, v. 165, no. 5, 1965, 1056-1058  TOPIC TAGS: giant pulse laser, dysprosium, calcium fluoride, xenon lamp, pure colcium fluoride, crystal, from purpose, flower fluoride, crystal, flower purpose, flower fluoride, crystal, flower purpose, flower fluoride, crystal, flower purpose, flower fluoride, xenon lamp, purposed for the pumped continuously by xenon lamps. Such pulses were first achieved in CaF2 by Ye. M. Zolotov, A., M. Prokhorov, and G. P. Shipulo (ZhETF, v. 49, no. 9, 1965), who used ruby laser pumping. A similar method of generating giant pur YALG:Nd was used by J. E. Gausic, M. L. Hensel, and R. G. Smith (Appl. Phys. 6, no. 9, 175, 1965). The laser system used in the present investigation (Flooristed of a cylindrical dysprosium-doped calcium fluoride crystal 70 mm le	CaF <sub>2</sub> :Dy <sup>2+</sup> :Dy <sup>2+</sup> 720, lses in Lett., ig. 1)
7 mm in diameter with plane-parallel ends. The concentration of $Dy^{2+}$ in CaF $\sim 10^{17}$ cm <sup>-3</sup> . The crystal was placed in a dewar, where it was cooled by circuliquid nitrogen. The pumping was provided by two cw xenon lamps placed togeth a dewar in a tight condenser. An internal multilayer dielectric mirror with	2 was lating ther with
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ACC NR: AP6002423

flectivity of approximately 100% was used on one end of the resonator, whose output was Q-switched by means of a rotating (50-500 cps) prism with total internal reflection. The laser beam was incident (at 23°) at a plane-parallel quartz plate and directed at a calorimeter and a liquid-nitrogen-cooled InSb photodiode with a time-resolution of 20·10-9 sec. The time-dependent emission intensity was recorded by

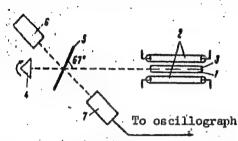


Fig. 1. Schematic of the laser system

1 - CaF<sub>2</sub>:Dy<sup>2+</sup> crystal; 2 - continuous pumping xenon lamps; 3 - multilayer dielectric mirror; 4 - rotating prism with total internal reflection; 5 - plane-parallel quartz plate; 6 - calorimeter; 7 - InSb photodiode.

means of an InSb photodiode and DEO-1 and S1-11 oscillographs. The mean intensity in both directions (see Fig. 1) was 0.05 w for both fixed and rotating (at 200 cps) prisms. This indicates that the rotation frequency of the prism was near optimal. The duration and repetition rate of the giant pulses were 1.2 x  $10^{-7}$  sec (calculated value was 1.05 x  $10^{-7}$ ) and 200 cps, respectively, resulting in a peak power of 2 x  $10^{3}$  w. The proposed high-intensity laser can be used in studies of two-photon excitation of semiconductors with a narrow forbidden gap. Orig. art. has: 2 figures.

SUB CODE: 20 / SUBM DATE: 02Sep65/ ORIG REF: 003/ OTH REF: 005/ ATD PRESS:

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37 no.1:53-54 Ja '60.	waginitis and vestibulitis.	(MIRA 16:6)
l. Kazanekiy nauchno-issledo (Propolis) (V	watel'skiy veterinarnyy inst Yaginitis in cattle)	itut.
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# Malignant neoplasms in farm animals. Veterinariia 38 no.6:59 Je '61. (MIRA 16:6) 1. Masanekaya veterinarnaya poliklinika. (Cancer) (Veterinary medicine)

KULEY W, Y. G.

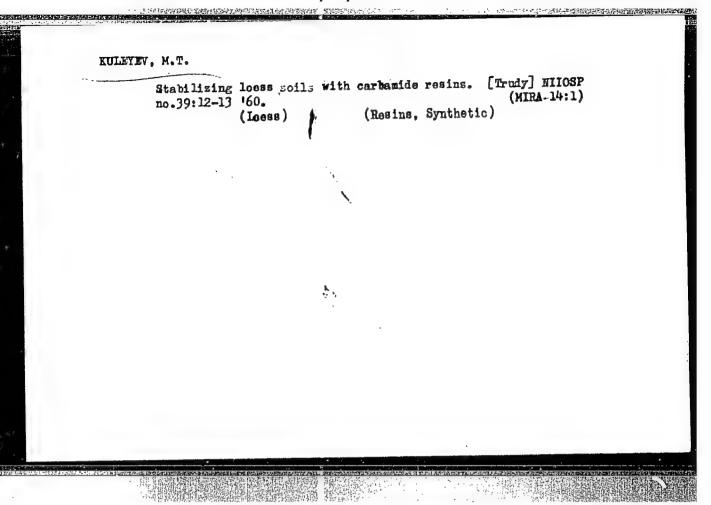
Kuleyev, I. G. "Transitory operation of motors with special rotors", Izvestiya Tomskogo pelitekhn. in-ta im. Kirova, Vol. LXIV, Issue 1, 1948, p. 29-36

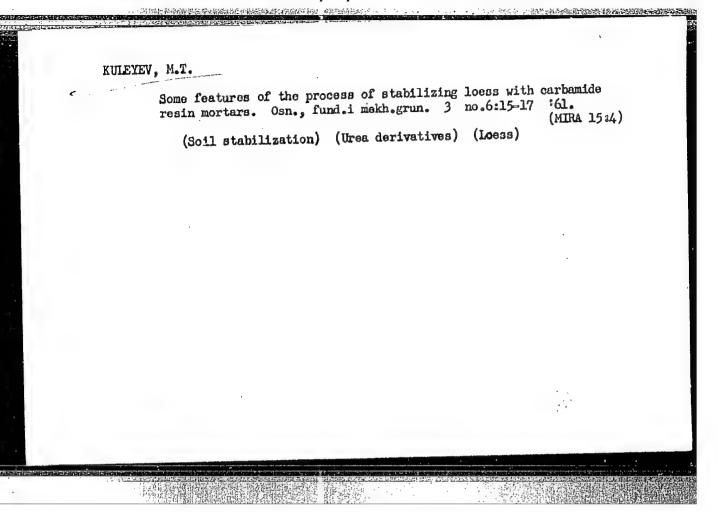
SO: U-4631, 16 Sept. 1953, (Letopis 'Zhurnal 'nykh Statey, No. 24, 1949)

KULTYTV, I. G.

Kuleyev, I. G. "A new method of reducint the curve of dependence of the moment on the number of windings in a shortcircuited [shunt-wond?] motor", Izvestiya Tomskogo politekhn. in-ta im. Kirova, Vol. LEIV, Issue 1, 1948, p. 37-39

SO: U-4631, 16 Sept. 53, (Lethopis 'Zhurnal 'nykh Statey, No. 24, 1949).





KULEYEV, M.T.

Strengthening loss soil with water solutions of carbamide resin in the field. [Trudy] NII osn. no. 50:10-14 '62. (MIRA 16:9)

EMT(1)/EPA(s)-2/EMT(m)/EMP(w)/T TMP(t)/EMP(b) IJP(c) L 1985-66 UR/0056/65/049/001/0248/0256 ACCESSION NR: AP5019238 Kuleyev, V. G AUTHOR: Turov. Ye. A.: TITLE: On coupled oscillations of electronic and nuclear spins in antiferromagnets Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 49, no. 1, 1965, SOURCE: 248-256 TOPIC TAGS: antiferromagnetism, electron spin, nuclear spin, spin wave ABSTRACT: The spectrum of coupled oscillations of electronic and nuclear spins (the coupling is due to hyperfine interaction) is calculated and studied in an antiferromagnet in which the axis of antiferromagnetism lies in a plane with small magnetic anisotropy. A mechanism is considered for the relaxation of the oscillations of the nuclear-like branch, which appears because of the hyperfine coupling

when account is taken of damping in the electronic spin system. The microwave magnetic susceptibility of the whole spin system is calculated, and the amplification coefficient for nuclear magnetic resonance is found. The spatial dispersion of nuclear-like spin waves is studied, and it is pointed out that such waves can be excited by a uniform microwave field (nuclear spin-wave resonance). Orig. art.

Card 1/2

has: 1 figure and 37 formulas.

L 1985-66 ACCESSION NR: AP5019238						• •	3	
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ASSOCIATION: Institut fiziki metallov Akademii nauk SSSR (Institute of Metal Physics, Academy of Sciences, SSSR)								
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. KULEZNEY, V.N.

69-20-1-7/20

AUTHORS:

Dogadkin, B.A., Kuleznev, V.A., Tarasova, Z.N.

TITLE:

Formation and Properties of Interpolymers of Natural and Butadiene-Styrene Rubbers (Polucheniye i svoystva mezhpolimerov natural'nogo i butadienstirol'nogo kauchukov)

PERIODICAL: Kolloidnyy Zhurnal, 1958, Vol. XX, # 1, pp 43-51 (USSR)

ABSTRACT:

The coplastication of natural and butadiens-styrene rubbers by milling on a cold mill leads to the formation of an interpolymer containing 30% of the natural rubber introduced. The plastication was carried out on a specially constructed micromill in a hermetic casing. The milling was done in an atmosphere of purified nitrogen. The rubbers were preliminarily purified by hot acetone (natural rubber) or hot methanol (Butadiene-styrene rubber). The values for the characteristic viscosity and plasticity during milling are represented in figures 1 and 2. To prove the formation of an interpolymer during milling, several methods were used. In one, fractional precipitation, a selective precipitator had to be found; used was a binary mixture (1:4) of benzene-methylethylketone, in which

Card 1/4

69-20-1-7/20

Formation and Properties of Interpolymers of Natural and Butadiene Styrene Rubbers

butadiene-styrene rubber dissolves completely, whereas natural rubber does not dissolve. For comparison the milled polymers were also dissolved. The solution was then separated, evaporated, and the content of natural rubber determined by an Abbe refractometer. Fig. 3 shows that in case of separately milled polymers the natural rubber begins to dissolve after 40 min. For selective vulcanization, polychloro-compounds were used, which do not vulcanize natural rubber. As an activator, ZnO and PbO in two parallel experiments was employed. The results have shown that 20-26% of the introduced natural rubber is being bound during plastication. The characteristic viscosity depends on the ratio of the rubbers in the mixture. Fig. 5 shows, that the values for the viscosity of the jointly milled polymers are higher than the corresponding values of the separately milled polymers. The investigation of the physical-chemical properties of the vulcanizates shows that the mixtures of natural and butadiene-styrene rubbers have a higher resistance to aging than natural rubber alone. The resistance to breaking, relative stretching and deformation is also dependent on the composition

Card 2/4

69-20-1-7/20

Formation and Properties of Interpolymers of Natural and Butadiene Styrene Rubbers

of the mixture. An adhesive film made from interpolymers increases the binding strength between natural and butadiene - styrene rubbers when placed between them. It is supposed that the segments of the molecules of the natural rubber in the interpolymer, which are connected with the butadiene-styrene rubber by chemical bonds, penetrate easily into the natural rubber. The same is true for the segments of the butadiene-styrene rubber of the interpolymer, which penetrate into the butadiene-styrene rubber. The results of the tests for resistance of the connections by interpolymer adhesive films are shown in table 2.

Card 3/4

There are 9 figures, 2 tables, and 7 references, 3 of which are Soviet, 4 English.

69-20-1-7/20

Formation and Properties of Interpolymers of Natural and Butadiene Styrene Rubbers

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii imeni M.V. Lomonosova (Moscow Institute of Fine Chemical Technology

imeni M.V. Lomonosov). Nauchno-issledovatel'skiy institut shinnoy promyshlennosti (Scientific Research Institute of

the Tire Industry)

SUBMITTED: July 12, 1957

AVAILABLE: Library of Congress

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11、包含在各种中或自然的结构的影響的影響的影響的影響。

AUTHORS: Dogadkin, B.A., Kuleznev, V.H. SOV-69-20-5-21/23 TITLE: The Formation of a Gel in the Plastication of Natural Rubber and Its Effect on the Strength of Vulcanizates (Obrazovaniye gelya pri plastikatsii natural'nogo kauchuka i yego vliyaniye na prochnost: vulkanizatov). PERIODICAL: Kolloidnyy zhurnel, 1958, Vol XX, Nr 5, pp 674-675 (USSR) Gel formation has been observed during the plastication of ABSTRACT: natural rubber. In the early stages of plastication (10 min), a gel forms amounting to a maximum of 20%. Further rolling causes a mechanical dispersion of this gel fraction. If the argon atmosphere contains more than 0.1% oxygen, the free radicals are stabilized and no gel is formed. In the Card 1/2 absence of oxygen, ramified molecules appear which decrease

307-69-20-5-21/23

The Formation of a Gel in the Plastication of Natural Rubber and Its Effect on the Strength of Vulcanizates

the rupture resistance of the rubber to 20-40 kg/cm $^2$  in comparison to the usual values of 200-250 kg/cm $^2$ . There is 1 graph and 4 references, 1 of which is Soviet and 3

English.

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii im.

M.V. Lomonosova (Moscow Institute of Fine Chemical Techno-

logy imeni M.V. Lomonosov)

SUBMITTED: June 10, 1958

1. Gels--Development 2. Rubber--Processing 3. Vulcanizates

--Mechanical properties

Card 2/2

KULEZNEV, V. N., Candidate Chem Sci (diss) -- "The preparation and properties of inter-polymers of natural and budadiene-styrol rubbers". Moscow, 1959. 13 pp (Min Higher Educ USSR, Moscow Inst of Fine Chem Tooh im M. V. Lemonosev), 150 copies (KL, No 24, 1959, 128)

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5(3) 504/69-21-2-9/22

AUTHORS: Dogadkin, B.A., Kuleznev, V.N., Pryakhina, S.F.

TITLE: On the Compatibility of Polymers in Solution (K voprosu

o sovmestimosti polimerov v rastvore)

PERIODICAL: Kolloidnyy ahurnal, 1959, Nr 2, pp 174-180 (USSR)

ABSTRACT: This is a report on an investigation concerning the behaviour of mixtures of natural and butadiene styrene rubber

in a common solution. The experiments have shown that mixtures of 5% benzene solutions of natural and butadiene styrene rubber exfoliate, if these substances are mixed within the limits 1:9 and 9:1. The concentration of the laminae is not equal to the initial concentration. For any ratio of rubber mixtures, the experimental viscosity values are higher, whereas the turbidity of the solutions is lower than the additive magnitudes. The increase in temperature, or the introduction of large quantities of

methyl ethyl ketone bring together the experimental and additive values of the viscosity of the solutions, i.e.

Card 1/2 they increase the compatibility of natural and butadiene

304/69#21-2-9/22

On the Compatibility of Polymers in Solution

styrene rubber. The presence of an interpolymer in a natural and butadiene styrene rubber mixture prevents exfoliation of the solution. According to the authors the observed phenomena may be considered as the result of molecular associations of prevalently homogeneous composition. There are 5 graphs and 10 references, 6 of which are Soviet and 4 English.

ASSOCIATION:

Moskovskiy institut tonkoy khimicheskoy tekhnologii im.M.V. Lomonosova (Moscow Institute of Fine Chemical Technology imeni M.V. Lomonosov)

SUBMITTED:

May 9, 1958

Card 2/2

STATISTICAL CONTROL OF STATISTICS OF STATIST

S/069/62/024/003/004/006 B110/B138

AUTHORS:

Kuleznev, V. N., Igosheva, K. M.

TITLE:

Effect of various substances on the stability of mixed polymer-

solutions

PMRIODICAL: Kolloidnyy zhurnal, v. 24, no. 3, 1962, 306 - 308

TEXT: An attempt was made to decelerate the separation of polymer mixtures by adding small amounts of polar substances. Solutions of the following technical, nonfractionated polymers were studied: block polystyrene and emulsion polymethyl methacrylate in cryoscopic benzene (polymer ratio=1:1), the mixtures of which separate at 9%. The following additives were used: Propyl, amyl, and ethyl alcohols, acetone, acetophenone, benzophenone, methyl-ethyl ketone, butyric, isobutyric and oleic acids, butyl acetate, benzyl acetate, methyl, ethyl, butyl, and isoamyl benzoates, aniline, dimethyl uniline, dichloro ethane, chlorobenzene, ethylene chlorohydrin, pyridine, and thiophen. Aniline, and ethylene chlorohydrin increased the optical density and accelerated the separation. In 10% solutions of mixtures (polystyrene: polymethyl methacrylate = 1:1) with 100 mole% substance per Card 1/2



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Effect of various substances ...

mole polymethyl methacrylate, the separation was only inhibited by dimethyl aniline and heptyl and amal alcohols. Benzoic acid ester, pyridine, dichloro ethane, acetophenone, methyl-ethyl ketone, benzophenone, thiophen, and butyl acetate decelerated the separation (induction period: 8 - 40 hrs). Separation became faster with increasing optical density due to admixtures. No relation was found between the effect of additive and surface tension, dipole moment, the dielectric constant and the heat of vaporization. There are 3 figures and 1 table.

ABSOCIATION: Ural'skiy gosudarstvennyy universitet im. A. M. Gor'kogo

(Ural State University imeni A. M. Gor'kiy)

SUBMITTED: July 20, 1961

Cará 2/2

\$/069/62/024/005/010/010 B117/B186

AUTHORS:

Kuleznev, V. N., Dogadkin, B. A.

TITLE:

Dependence of the mechanical properties of a polymer mixture

on its composition

PERIODICAL:

Kolloidnyy zhurnal, v. 24, no. 5, 1962, 632-633

TEXT: Experiments upon natural and butadiene styrene rubber have shown that the properties of specially prepared homogeneous mixtures of thermodynamically incompatible polymers are not of an additive nature. Whatever the technique of preparation (rolling in argon or mixing in air and gamma irradiation), intermediate polymers were formed in mixtures that had been vulcanized at 142°C under pressure. Curves showing the mechanical properties of 1:1 mixtures versus their composition were characterized by extreme values (increase in fatigue strength and decrease in ultimate strength). As such a mixture, wherein most of the macromolecules of one polymer are surrounded by the macromolecules of the other, does not form a crystalline phase when stretched, its ultimate strength is equal to or even lower than that of non-crystallizing rubber. The increase in dynamic

Card 1/2

Dependence of the mechanical ...

S/069/62/024/005/010/010 B117/B186

strength has so far not been explained. There is 1 figure.

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii imeni M. V. Lomonosova (Moscow Institute of Fine Chemical

Technology imeni M. V. Lomonosov)

SUBMITTED:

May 22, 1962

Card 2/2

KULEZNEV, V.N.; ANDREYEVA, V.M.

Light scattering by solutions of polymer mixtures. Vysokom. soed. 4 no.12:1851-1857 D '62. (MIRA 15:12)

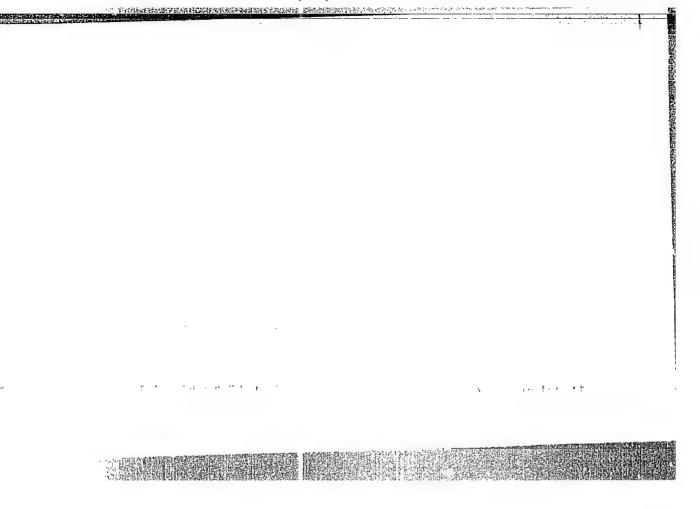
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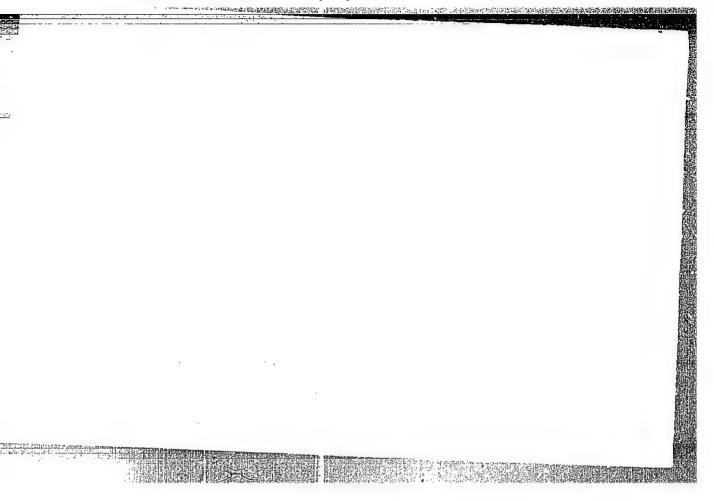
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KULEZNEV, V.N.; IGOSHEVA, K.M.

Densities of polymer mixtures. Vysokom. soed. 4 no.12:1858-1862 D 162. (MIRA 15:12)





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ACCESSION NR: AP4043129

3/0069/64/026/004/0475/0480

AUTHORS: Kuleznev, V.N.; Krokhina, L.S.; Lyakin, Yu.I.; Dogadkin, B.A.

TITLE: Investigation of the structure of solutions of polymer mixtures by the light scattering method

SOURCE: Kolloidny\*y zhurnal, v. 26, no. 4, 1964, 475-480

TOPIC TAGS: light scattering, polystyrene polyisobutylene toluene system, refractive index, polymer association, apparent molecular weight, polystyrene, polystyrene association, asymmetry of light scattering, second virial coefficient, true solution, emulsion, colloid

ABSTRACT: This study of light scattering (at  $R_{15}^{m}$ ,  $R_{0}^{m}$ , and  $R_{135}^{m}$ ) in the polystyrene-polyisobutylene-toluene system in which the refractive indices of the polyisobutylene (M = 1.2 x 10°) and toluene are equal provided new proof of the increased degree of association of one polymer upon the addition of a second polymer. Optical densities of the solutions illuminated with monochromatic light of 5460 and 4360A were measured on the FEKN-56 photoelectric nephelometer. The association of polystyrene, i.e., the apparent molecular

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ACCESSION NR: AP4043129

weight, and the asymmetry of the scattering increased and the second virial coefficient diminished on addition of the optically invisible polyisobutylene. Quantitative measurements of the degree of association are possible only on double extrapolation of the light scattering to zero angle and zero concentration. On changing in concentration a mixture of two polymers may pass from a true solution to a rapidly separating emulsion via a stable intermediate colloid state. For a 1:1 mixture of polystyrene and polyisobutylene the true solution exists up to a concentration of 0.7 gm. of the mixed polymers per 100 ml. toluene, and the emulsion separates when the concentration exceeds about 1.2 gm./100 ml. "The authors sincerely thank V.E. Eskin for participation in evaluating the results and I.M. Bel'govsk for permitting work on the photoelectric nephelometer." Orig. art. has: 5 figures and 1 table.

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii im. M.V. Lomonosova (Moscow Institute of Fine Chemical Technology)

\*Card 2/3

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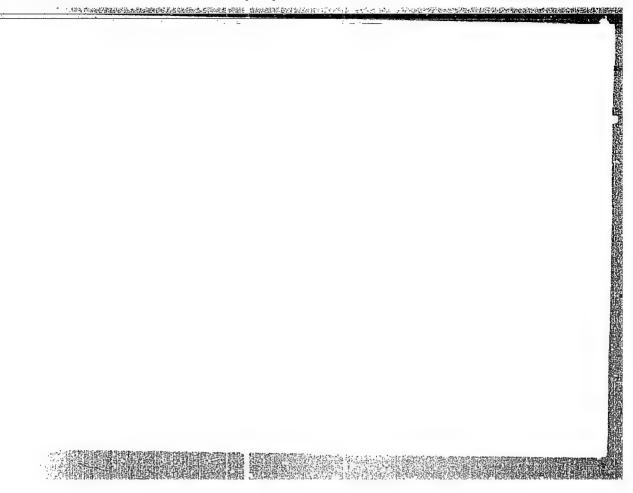
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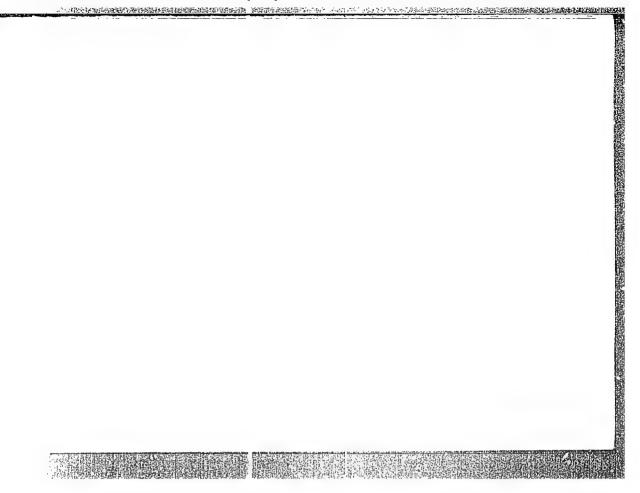
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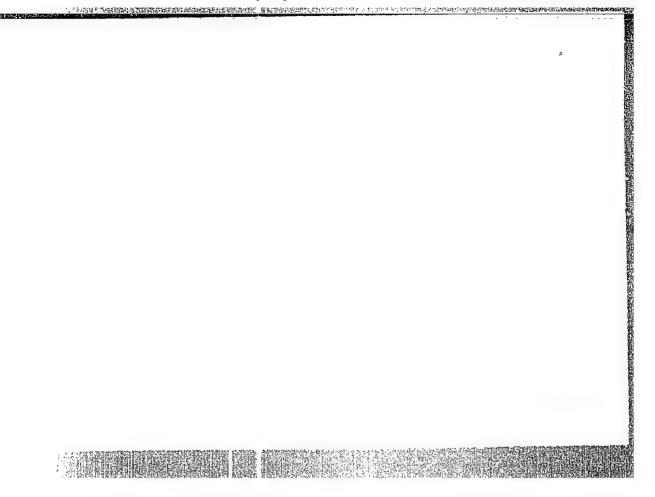
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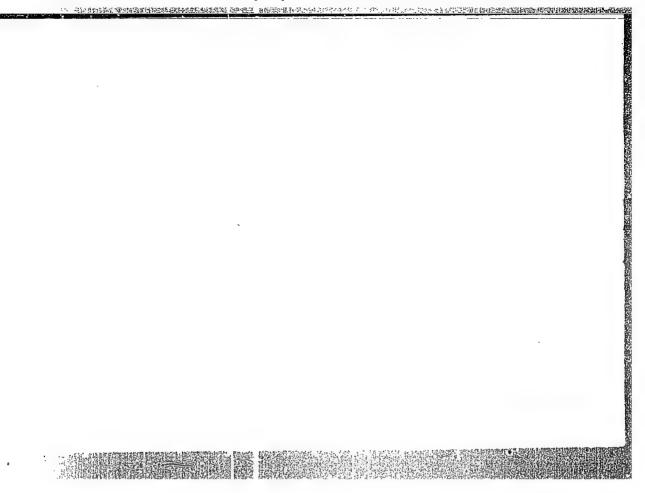
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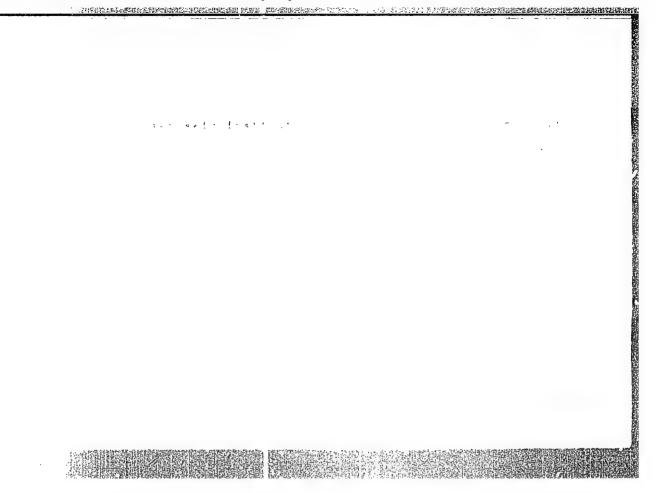
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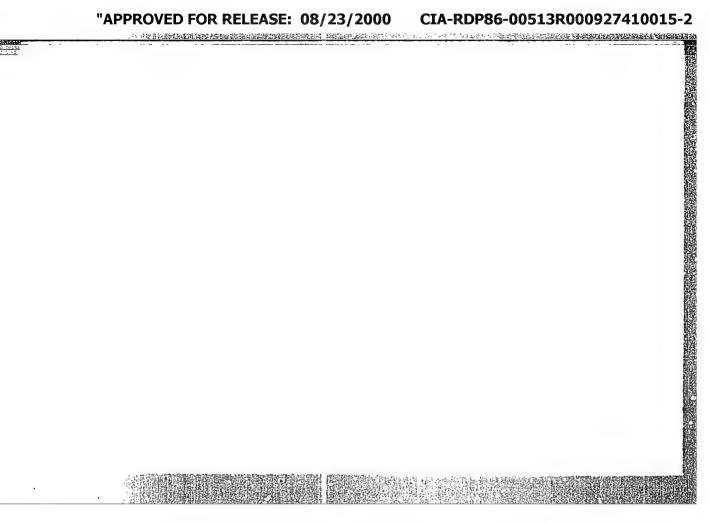


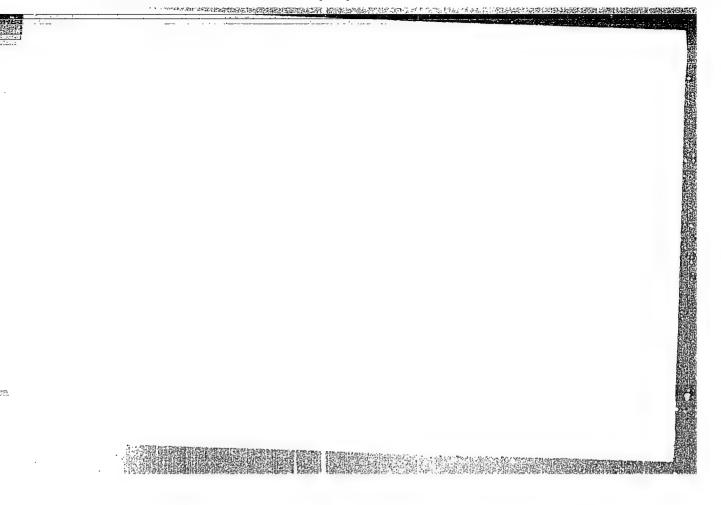




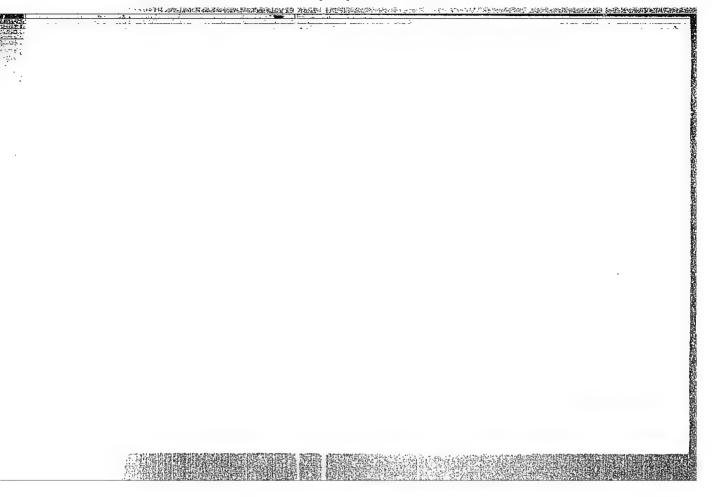


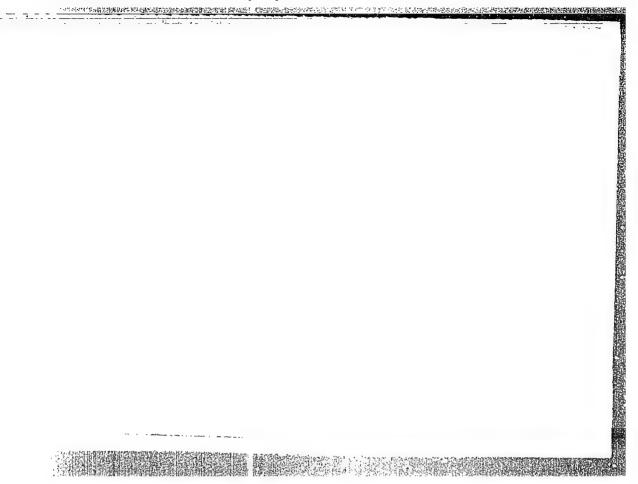


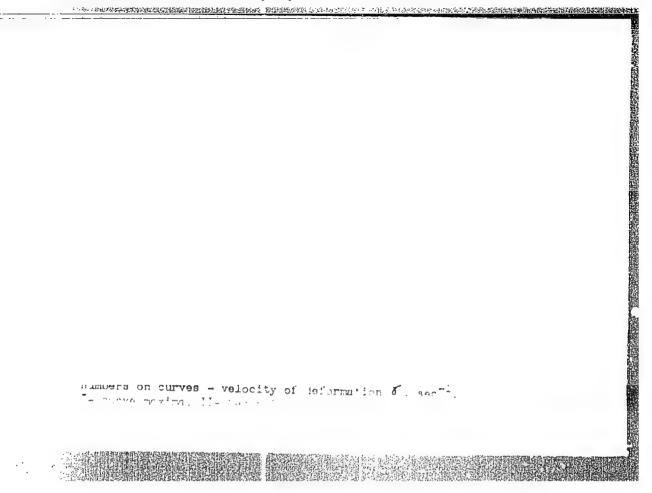


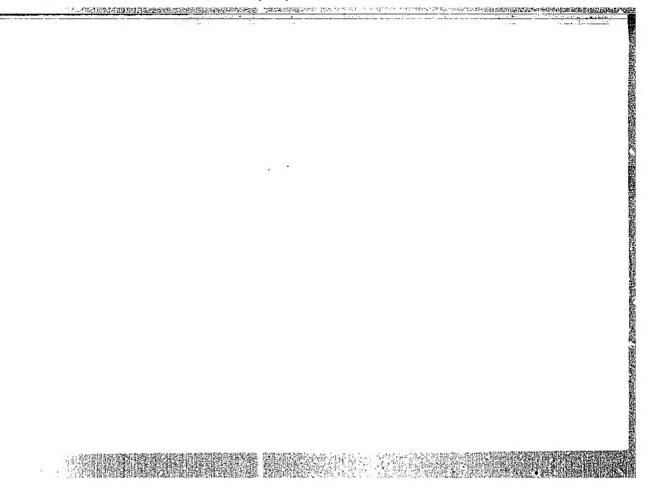


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KULEZNEV, V.N.; KROKHINA, L.S.; DOGADKIN, B.A.

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